# Functionalized Chromans and Isochromans via a Diastereoselective Pd(0)-Catalyzed Carboiodination

### David A. Petrone, Hasnain A. Malik, Antonin Clemenceau, and Mark Lautens\*

Department of Chemistry, Davenport Chemical Laboratories, 80 St. George Street, University of Toronto, Toronto, Ontario M5S 3H6, Canada

mlautens@chem.utoronto.ca

#### Received July 31, 2012



A diastereoselective approach to isochromans and chromans via Pd(0)-catalyzed carboiodination is reported. The transformations using this methodology display excellent yields and diastereoselectivities as well as broad functional group compatibility. The selectivity observed in these cyclizations, forming isochroman or chroman targets, is postulated to originate from the minimization of  $A^{1,2}$  strain and axial—axial interactions, respectively. This method has also been used to highlight the concept of reversible oxidative addition to carbon-iodine bonds in polyiodinated substrates.

Heterocycles appear in numerous bioactive compounds and important synthetic intermediates.<sup>1</sup> The continued development of efficient strategies which allow the synthesis of these heterocyclic frameworks is of great importance to the synthetic community.<sup>2</sup> Transition-metal catalyzed methods resulting in newly formed carbon-carbon bonds have emerged as an effective strategy toward this end.3 Furthermore, there is a continuing need for the development of atom-economical variants of organometallic transformations that result in the retention of important functionality within a molecule. Our group<sup>4a-c</sup> and others<sup>4d,e</sup> are interested in using  $Pd(0)$ -catalyzed carbohalogenation<sup>5</sup> as an efficient and atom-economical<sup>6</sup> synthetic strategy toward complex ring systems. Recently, we reported a Pd(0)-catalyzed bromide to iodide exchange reaction which was incorporated within various domino cyclizations.<sup>7</sup> These

## ORGANIC **LETTERS** 2012

Vol. 14, No. 18 4806–4809

<sup>(1) (</sup>a) Majumdar, K. C.; Chattopadhyay, S. K. Heterocycles in Natural Product Synthesis; VCH: New York, 2011.

<sup>(2)</sup> Bolm, C.; Beller, M. Transition Metals for Organic Synthesis; VCH: New York, 2004; Vols. 1 and 2.

<sup>(3) (</sup>a) Zeni, G.; Larock, R. C. Chem. Rev. 2004, 104, 2285–2309. (b) Zeni, G.; Larock, R. C. Chem. Rev. 2006, 106, 4644-4680. (c) Li, J. J.; Gribble, G. W. Palladium in Heterocyclic Chemistry; Pergamon: New York, 2000. (d) Negishi, E. Handbook of Organopalladium Chemistry for Organic Synthesis; Wiley and Sons: New York, 2002; Vols. 1 and 2. (e) Fagnou, K.; Lautens, M. Chem. Rev. 2003, 103, 169–196.

<sup>(4) (</sup>a) Newman, S. G.; Lautens, M. J. Am. Chem. Soc. 2010, 132, 11416–11417. (b) Newman, S. G.; Lautens, M. J. Am. Chem. Soc. 2011, 133, 1778–1780. (c) Lan, Y.; Liu, P.; Newman, S. G.; Lautens,M.; Houk, K. N. Chem. Sci. 2012, 3, 1987–1995. (d) Liu, H.; Li, C.; Qiu, D.; Tong, X. J. Am. Chem. Soc. 2011, 133, 6187–6193. (e) Liu, H.; Chen, C.; Wang, L.; Tong, X. Org. Lett. 2011, 13, 5072–5075.

<sup>(5)</sup> For the seminal report of C-I reductive elimination using rhodium, see: Feller, M.; Iron, M. A.; Shimon, L. J. W.; Diskin-Posner, Y.; Leitus, G.; Milstein, D. J. Am. Chem. Soc. 2008, 130, 14374–14375.

<sup>(6) (</sup>a) Trost, B. M. Science 1991, 254, 1471–1477. (b) Trost, B. M. Angew. Chem., Int. Ed. Engl. 1995, 34, 259–281.

<sup>(7)</sup> Newman, S. G.; Howell, J. K.; Nicolaus, N.; Lautens, M. J. Am. Chem. Soc. 2011, 133, 14916–14919.

reactions afforded a diverse set of complex polycyclic products in good yields with moderate to excellent levels of diastereocontrol. We were interested in exploring levels of diastereocontrol in this reaction for the synthesis of chroman and isochroman motifs, as their derivatives display interesting biological and medicinal activity.8

Recent synthetic strategies toward their synthesis include the use of Lewis acids, $9$  organocatalysis, $10$  and palladium catalysis,  $^{11}$  among others.<sup>12</sup> Despite these methods affording noteworthy reactivity and stereoselectivity, there is a lack of general and unified approaches toward both these heterocyclic skeletons.13 We envisioned utilizing our palladium catalyzed carboiodination reaction as anintegrated approach to access both of these molecular frameworks. Herein, we report the synthesis of various functionalized chromans and isochromans via a highly diastereoselective intramolecular Pd(0)-catalyzed carboiodination of alkenyl aryl iodides.

Our investigation began by analyzing the effectiveness of various bulky phosphine-containing Pd(0) precatalysts on the intramolecular cyclization of 1a (Table 1). In accordance with previous reports,  $6b$ ,7 both Pd[(PCy<sub>3</sub>)]<sub>2</sub> (entry 1) and  $Pd[P(o-tol)_3]_2$  (entry 2) gave no desired isochroman product. The more bulky ligand<sup>14</sup>  $P<sup>t</sup>Bu<sub>2</sub>Ph$  only afforded trace conversion to the desired product 2a (entry 3) despite performing moderately well in previous systems. Q-Phos was much better, affording 62% of the desired product with a diastereomeric ratio of 90:10 (entry 4). When 5 mol  $\%$  P'Bu<sub>3</sub><sup>15</sup> was employed, both yield and diastereoselectivity increased to 94% and 94:6, respectively (entry 5). In the absence of  $NEt_3$  we noticed a significant decrease in yield (47%) but no decrease in stereoselectivity (entry 6). Decreasing the catalyst loading to 2.5 mol  $\%$ caused a marked decrease in overall yield (77%, entry 7). The optimal reaction conditions were found to be 5.0 mol  $\%$  $Pd[(P^TBu_3)]_2$ , 1 equiv of NEt<sub>3</sub> in toluene at 110 °C. The requirement of an amine base is postulated to assist the regeneration of Pd(0) from Pd(II)HX, which may result from trace intermolecular Mizoroki-Heck type processes.

(8) Ellis, G. P.; Lockhart, I. M. The Chemistry of Heterocyclic Compounds, Chromenes, Chromanones, and Chromones; VCH: New York, 2007.

(9) van Lingen, H. L.; Zhuang, W.; Hansen, T.; Rutjes, F. P. J. T.; Jorgensen, K. A. Org. Biomol. Chem. 2003, 1, 1953–1958.

 $(10)$  (a) Lee, Y.; Seo, S. W.; Kim, S. G. Adv. Synth. Catal. 2011, 353, 2671–2675. (b) Yu, D. F.; Wang, Y.; Xu, P. F. Adv. Synth. Catal. 2011, 353, 2960–2965. (c) Enders, D.; Urbanietz, G.; Hahn, R.; Raabe, G. Synthesis 2012, 44, 773. (d) Hu, F.; Guan, X.; Shi, M. Tetrahedron 2012, 68, 4782–4790.

(11) (a) Yu, L.; Wang, D.-H.; Engle, K. M.; Yu, J.-Q. J. Am. Chem. Soc. 2010, 132, 5916–5921. (b) Leibeling, M.; Milde, B.; Kratzert, D.; Stalke, D.; Werz, D. B. Chem.--Eur. J. 2011, 17, 9888-9892. (c) Ward, A. F.; Xu, Y.; Wolfe, J. P. Chem. Commun. 2012, 48, 609–611. (d) Cannon, J. S.; Olsen, A. C.; Overman, L. E.; Solomon, N. S. J. Org. Chem. 2012, 77, 1961–1973.

(12) (a) Medeiros, M. R.; Narayan, R. S.; McDougal, N. T.; Schaus, S. E.; Porco, J. A., Jr. Org. Lett. 2010, 12, 3222–3225. (b) Kern, N.; Blanc, A.; Weibel, J. M.; Pale, P. Chem. Commun. 2011, 47, 6665–6667. (c) Shen, H. C. Tetrahedron 2009, 65, 3931–3952.

(13) Leibeling, M.; Koester, D. C.; Pawliczek, M.; Schild, S. V.; Werz, D. B. Nat. Chem. Biol. 2010, 6, 2010.

(14) Tolman, C. A. Chem. Rev. 1977, 77, 313–348.

(15) For syntheis of this ligand, see: (a) Li, H.; Grasa, G. A.; Colacot, T. J. Org. Lett. 2010, 11, 3332–3335. (b) Colacot, T. J.; Grasa, G. A.; Li H. B. (Johnson Matthey Public Ltd. Co., USA) Preparation of a metal complex. International Patent WO 2010/128316 A1, November 11, 2010.

#### Table 1. Reaction Optimization





 $\alpha$  Calculated by  $\rm ^1H$  NMR analysis of the crude reaction mixture using 1,3,5-trimethoxybenzene as an internal standard.  $\frac{b}{b}$  Value in brackets represents yield of unreacted starting material.  $\cdot c$  Calculated by  $\cdot^1$ H NMR analysis of the crude reaction mixture. <sup>d</sup>Isolated yield. <sup>e</sup> Reaction run in the absence of NEt<sub>3</sub>.

Table 2. Scope of Isochroman Synthesis<sup>a</sup>



 $a$  Reaction conditions: Aryl iodide (0.2 mmol, 0.05 M), NEt<sub>3</sub> (0.2 mmol)  $Pd(P'Bu_3)_2$  (5 mol %), toluene. <sup>b</sup> Isolated yield. <sup>c</sup> dr calculated by  ${}^{1}H$  NMR analysis of the crude reaction mixtures.  ${}^{d}$  Reaction run at  $0.1$  M with respect to the aryl iodide.  $e^{i}$  Yield obtained when 100 mol % of pyridine was added to the reaction mixture.  $f_{1.0}$  equiv of  $iPr_2$ NEt was used instead of  $NEt<sub>3</sub>$ .

A diverse set of substituted alkenyl aryl iodides were subjected to the reaction conditions to test the scope and selectivity of this transformation (Table 2). When the cyclization of 1a was conducted on a 2 mmol scale, the

Table 3. Scope of Chroman Synthesis<sup>a</sup>



<sup>*a*</sup> Reaction conditions: Aryl iodide (0.2 mmol, 0.1 M), NEt<sub>3</sub> (0.2 mmol),  $Pd(P'Bu<sub>3</sub>)<sub>2</sub>$  (5 mol %), toluene. <sup>b</sup> Isolated yield. <sup>c</sup> Calculated by <sup>1</sup>H NMR analysis of the crude reaction mixtures.



Figure 1. X-ray structures of 2b and 4c.

corresponding isochroman was afforded in 88% yield with no erosion of diastereoselectivity. Benzodioxole substituted 2b was afforded in 91% yield with 96:4 diastereoselectivity. The *cis* isomer was unambiguously identified as the major product by X-ray crystallographic analysis (Figure 1a). This stereoselectivity of isochroman formation is thought to arise from the decreased  $A^{1,2}$  strain<sup>16</sup> of the tether substituent and the vinylic  $CH_3$  prior to carbopalladation (Scheme 1a).

Heterocyclic substituents were tolerated affording the desired isochromans 2c and 2d in 85% and 94% yields with 92:8 and 91:9 diastereoselectivity, respectively. 3-Pyridyl substituted 2f was obtained in 35% yield with 98:2 diastereoselectivity. However, the 2-pyridyl analog did not undergo the desired transformation, providing only recovered starting material. Cyclohexyl containing 2g was isolated in Scheme 1. Stereochemical Models Describing the Diastereoselective Carbopalladation Step for (a) Isochroman and (b) Chroman Formation



81% yield and 91:9 diastereoselectivity, while cyclopropyl 2h and *n*-propyl-containing 2i analogs were obtained with yields of 93% and 86% and diastereoselectivities of 90:10 and 89:11, respectively. This general trend of stereoselectivity is dependent on overall steric factors in order to obtain high levels of diastereoselectivity. As evidence of these steric requirements,  $o$ -CF<sub>3</sub> functionalized 2j was obtained in 88% yield as a single observable diastereomer. It is noteworthy that an enantiomerically enriched precursor (S)-1k was prepared and cyclized to the corresponding isochroman  $(3R,4R)$ -2k with high levels of diastereoselectivity and no apparent erosion of enantiomeric excess (eq 1).



This methodology was further applied to the synthesis of substituted chromans (Table 3). 2-Phenyl chroman 4a was obtained in 81% yield and 93:7 diastereoselectivity (entry 1). A 1-napthalene analog 4b was afforded in 96% yield as a single diastereomer (entry 2). Notably, a switch in stereochemistry occurs from *cis* to *trans* with respect

<sup>(16) (</sup>a) Hoffman, R. W. Chem. Rev. 1989, 89, 1841–1860. (b) JohnsonF. Chem. Rev. 1968, 68, 375–413.

to the isochroman variants, and this stereochemical outcome was unambiguously determined through X-ray analysis (Figure 1b). The observed stereoselectivity is thought to originate from the minimization of axial—axial interactions in the carbopalladation step (Scheme 1). This reaction is exceptionally tolerant toward heteroatoms, halogens, and electron-withdrawing groups, as the desired chromans were obtained with excellent yields and diastereoselectivities  $(4c-e)$ .

Remarkably, di-iodinated substrate 3f was converted to the corresponding chroman 4f in 70% yield and 91:9 diastereoselectivity without deleterious byproduct formation (eq 2). This result is consistent with an earlier report that highlights the possibility of reversible oxidative addition to  $C-I$  bonds by the Pd-catalyst, since it might be anticipated that the 4-iodo would react before the more hindered 2-iodo group.<sup>4a</sup>



In conclusion, we have developed a common approach for the diastereoselective synthesis of isochroman and chroman frameworks via a Pd(0)-catalyzed carboiodination. These transformations display generally good to excellent yields and high diastereoselectivities and have broad functional group compatibility. The stereochemistry observed in these cyclizations is postulated to originate from the minimization of  $A^{1,2}$  strain and axial—axial interactions. A stereochemical model has been presented. Studies to further understand the involvement of an amine base, as well as computational analyses to explore the energetics of these cyclizations, are currently underway in our laboratory and will be reported in due course.

Acknowledgment. We thank the Natural Sciences and Engineering Research Council of Canada (NSERC) and the University of Toronto for financial support. NSERC/ Merck-Frosst Canada is thanked for an Industrial Research Chair. D.A.P. thanks NSERC for a postgraduate scholarship (CGS-M). We thank Johnson Matthey Catalysis and Chiral Technologies for donating palladium catalysts, including Pd(P'Bu<sub>3</sub>)<sub>2</sub>, and Alan J. Lough (Chemistry Department, University of Toronto) for obtaining the crystal structures of 2b and 4c. David A. Candito (University of Toronto) is thanked for helpful discussions.

Supporting Information Available. Experimental procedures, spectral data for all new compounds, and crystallographic data (CIF). This material is available free of charge via the Internet at http://pubs.acs.org.

The authors declare no competing financial interest.